



Mini Review on different instrumental approaches applied to some selected drugs for COVID-19 treatment

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ABSTRACT

The quickest methods for battling the new coronavirus pandemic at the beginning of the COVID-19 outbreak were therapeutic treatments based on approved drugs. Although there are now many vaccines available, it will take time for the global vaccination program to take effect. Therefore, in addition to vaccination, repurposing currently available antiviral, antibiotic, and other types of medications has been suggested as a complementary medical strategy for COVID-19 infections. Analytical methods for assessing drug concentrations in biological fluids and pharmaceutical products are necessary due to the drugs' extensive clinical potential as well as their potential side effects. The following article was introduced by presenting a mini review of the different methods used for the quantitative determination of these drugs in order to encourage and facilitate collecting literature review for other researchers who will have studies related to the presented compounds. The review discussed techniques used for the determination of each drug quantitatively involving spectrophotometric (UV and visible), spectrofluorimetric, electrochemical and chromatographic (TLC and HPLC) methods either in a biological sample or in the pharmaceutical dosage forms. All methods included in the review are validated and constructed according to ICH guidelines.

1. Introduction

The COVID-19 pandemic, the worst public health emergency in the previous 100 years, struck the world in 2020. Every continent, except for Antarctica, has been hit by the coronavirus disease-2019 (COVID-19) pandemic in December 2019. COVID-19 is a contagious illness linked to SARS-CoV-2, a new coronavirus that causes the severe acute respiratory syndrome. Even though the world has survived many pandemics in the past, this one is a unique global health challenge that has changed how we live and is having a devastating socioeconomic impact on people worldwide [1]

SARS-primary CoV-2's peculiarity is its hazy symptoms, commonly mistaken for the flu and common cold [2]. The fact that COVID-19 is unpredictable makes it even more concerning. It is especially scary because it can have fatal outcomes, such as pneumonia and acute respiratory distress syndrome, or cause non-to-mild respiratory tract symptoms in the majority of patients. Thankfully, most individuals who contract the virus can recover with only supportive treatment. This unpredictability of the virus is one more reason why it is important to take the necessary precautions to prevent its spread.

Patients with chronic respiratory diseases, cardiovascular diseases, cancers, and immune deficiencies are susceptible to serious pathological complications and death [3]. The serious consequences led to a worldwide search for a virus-specific treatment that would be effective. The level of public mistrust toward such vaccinations is significant, despite the approval and introduction of numerous vaccine types recently. These studies showed that existing therapies could be effective against SARS-COV-2, offering hope for a quicker path to treatment. Furthermore, the results indicated that it is possible to repurpose existing drugs for the treatment of SARS-COV-2 [4].

The first recognized therapy for severe coronavirus disease is remdesivir (I) in 2019. It is a novel nucleoside analog with broad antiviral activity against RNA viruses, such as respiratory pathogens and the Ebolavirus [5]. It is an analog of adenosine and a monophosphoramidate prodrug. Remdesivir is broken down into its active form, which blocks viral RNA polymerase and avoids being checked by viral exonuclease, resulting in a reduction in the production of viral RNA. Remdesivir inhibits the nascent viral RNA of the Ebola virus through a delayed chain termination mechanism [6]. As a result of the immediate need for treatment, various experimental agents that already existed have been tried. As a result, to help control viral replication and the patient's general health, COVID-19 patients take multiple therapy drugs from various categories [7].

Since the start of the COVID-19 pandemic, scientists have concentrated on finding new uses for current antibiotics, antivirals, and anti-inflammatory medications. Being chemically derived from quinoline, the prodrug of chloroquine, fluoroquinolones are synthetic, broad-spectrum antimicrobial agents. Curiously, fluoroquinolones have been shown to have antiviral effects on the vaccinia virus, papovavirus, CMV, VZV, HSV-1, HSV-2, HCV, and HIV [8]. Fluoroquinolones such as lomefloxacin (II) were suggested to be used as adjuncts in treating patients who have COVID-19 considering their potential antiviral activity against SARS-CoV-2, along with their immunomodulatory properties,

favorable pharmacokinetics, and excellent safety profile [8]. According to coronavirus disease 2019 (COVID-19) treatment guidelines, dexamethasone can be used in the treatment protocol. Corticosteroids' ability to treat pneumonia caused by COVID-19 and other coronaviruses was demonstrated in recent studies that showed this to be true both in vitro and in vivo. Dexamethasone (III) at low doses could decrease mortality in patients with severe COVID-19 disease, but it did not affect the mortality rate in patients with a milder form of the disease [9].

There is an urgent need for straightforward and reliable bioanalytical methods for drugs quantification in the human plasma matrix in order to advance clinical research and high-throughput monitoring. Here, we provide an overview of methods of instrumental analysis for drugs used in COVID-19 treatment especially remdesivir, lomefloxacin and dexamethasone.

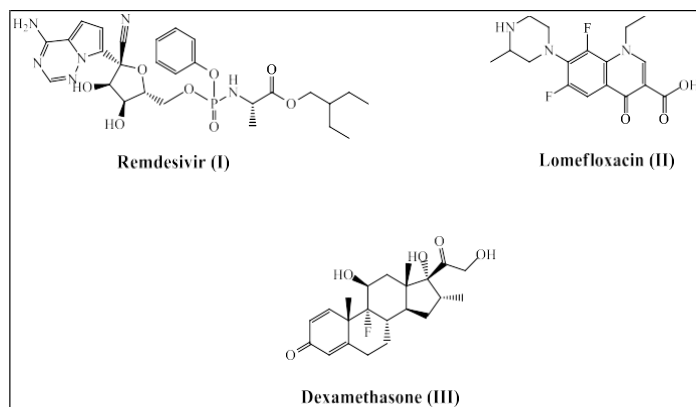


Figure 1: Chemical structure of remdesivir (I), lomefloxacin (II) and dexamethasone (III).

2. Methods of analysis of Selected drugs in this study Materials

2.1 Reported methods of remdesivir

2.1.1 Spectrophotometric methods

UV spectrophotometric methods have been reported for determination of remdesivir, including the following.

Formation of complex with acid dye bromophenol blue to produce a yellow ion-pair complex which can be measured at 418 nm. The best acid dye was chosen for this method by using computational and theoretical studies [10].

The synthesis of a novel charge transfer complex (CTC) between chloranilic acid (CLA), an electron acceptor, and REM, an electron donor, is described in this study for the first time. Different spectroscopic and thermal gravimetric methods were used to characterize the CTC. Through the development of a new broad absorption band with a maximum absorption peak (max) at 530 nm, UV-visible spectroscopy was demonstrated the formation of the CTC in methanol [11].

2.1.2 Spectrofluorimetric methods

Fluorescent spectroscopy has also garnered a lot of interest. Its benefits include environmental sustainability and analytical performance. Pharmaceutical quality control procedures must be delicate, quick, and economical to offer high throughput at a fair price [12]. Spectrofluorimetric methods have been reported for the determination of remdesivir, including the following:

One study was based on measurements of fluorescence (pH = 4) between 244 and 405 nm.

Calibration was completed over the 1.0 - 65.0 ng/mL range to improve sensitivity at detection and quantitation limits of 0.287 and 0.871 ng/mL, respectively. Other variables affecting this technique were also studied. [13]

The fluorescence intensity for remdesivir was recorded at λ emission (410 nm) after λ excitation at 241 nm [14].

2.1.3 Electrochemical technique

In this study, an anodic process utilizing a composite of Squaraine Dye and Ag₂O₂ has been assessed. The electro-analytical process has a branched mechanism, which suggests relatively dynamic behavior. However, the associated mathematical model analysis, conducted using the theories of linear stability and bifurcation, supports the composite electro-analytical efficiency as an electrode modifier [15].

2.1.4 Chromatographic methods

2.1.4.1. HPLC chromatographic methods

Table 1: HPLC methods reported for the determination of remdesivir.

| Matrix | Column | Mobile phase | Detector | Ref. |
|---|-------------------------------|--|--|-------------------------------------|
| Pharmaceutical formulations | An agilent Extend C18 | acetonitrile:KH ₂ PO ₄ solution (50:50, by volume) | UV at 247 nm | (Bulduk and Akbel 2021) [16] |
| Plasma with Abacavir (internal standard) | Chromosil C18 | Methanol:Acetonitrile: 0.1% OPA (30:65:5 v/v) | UV at 272 nm | (Kishore, Prasad et al. 2021) [17] |
| Spiked human plasma in the presence of frequently co-administered medications | Reversed phase agilent C18 | Gradient elution using (a) acetonitrile (b) water acidified at pH 4 with orthophosphoric acid. | Diode array detector at 240 nm Fluorescence λ_{ex} = 245 nm λ_{em} = 390 nm | (Moneim, Kamal et al. 2021) [7] |
| Rat plasma | Inertsil ODS column | Isocratic elution (a) Buffer of triethyl amine (b) Acetonitrile (50:50 v/v) | Triple quadrupole mass detector | (Rao, Adimulapu et al. 2022) [18] |
| Sublingual tablet dosage form | C18 column | Acetonitrile:Ammonium acetate buffer (pH 4.0) (40:60 % v/v) | Photo Diode Array Detector. | (Padhye, Sonawane et al. 2022) [19] |
| Rat plasma with Hydroxychloroquine -Favipiravir -Oseltamivir | C18 column | (a) Water (b) acetonitrile, (c) 0.1 % (v/v) formic acid | Triple Quadrupole MS | (El Azab 2022) [20] |
| Plasma with -Favipiravir -Dexamethasone | Reversed phase BEH C18 column | Acetonitrile:methanol:water acidified at (pH 4) with orthophosphoric acid (35:15:50, v/v) | Diode array detector | (Emam, Abdelaleem et al. 2022) [21] |

| | | | | |
|--|---------------------------|--|-------------------------------------|--------------------------------------|
| In human plasma with its active metabolite (GS441524) | Reversed phase agilentC18 | Acetonitrile:Dimethyl Sulfoxide (50:50 v/v) | Triple QuadrupoleMS | (Skaggs, Zimmerman et al. 2022) [22] |
| In human serum with chloroquine, Hydroxychloroquine, lopinavir, Ritonavir, favipiravir, azithromycin | Reversed phase C18 | (a) 0.1% formic acid (b) and acetonitrile | Triple QuadrupoleMS | (Habler, Brügel et al. 2021) [23] |
| In human plasma with its active metabolite (GS-44152) | Reversed phase agilentC18 | (a) formic acid (b) acetonitrile | Tandem quadrupole MS | (Kumar, Keerthana et al. 2022) [24] |
| In human plasma | Reversed phase C18 | (a) acetonitrile (b) 0.1% formic acid | Quadrupole tandem mass spectrometer | (Alvarez, Moine et al. 2020) [25] |

2.2 Reported methods of Lomefloxacin

2.2.1 Spectrophotometric methods

UV spectrophotometric methods have been reported for determination of lomefloxacin hydrochloride, including the following:

Formation of complex with praseodymium at pH 6.5–8.5 which can be measured using the second derivative spectra at 357 nm [26] (Wang, Ren et al. 2000). Measurement of the absorbance of the drug aqueous solution using distilled water at 280 nm [27]. First derivative for lomefloxacin determination in presence of its acid degradation product with zero crossing point at 295.2 nm [28].

Measuring the absorbance at the 287 nm using different media such as water, 0.1N HCl 0.1N NaOH and chloride buffer that used as solvent [29].

Measuring the absorbance at the 281 nm using urea solution (8M) as solubilizing agent [30].

Determination of the drug in presence of gemifloxacin mesylate and photodegradation products using two wavelengths 327 and 278 nm [31].

Measurement of the absorbance of the drug aqueous solution using distilled water at λ_{max} = 287.1 nm [32].

2.2.2 Spectrofluorimetric and chemiluminescence methods

One method depends on complex formation between the drug and terbium ion (Tb^{3+}) to enhance the

fluorescence intensity with λ_{ex} = 320 nm and λ_{em} = 545 nm [33]. Chemiluminescence method depends on that redox reaction between cerium (Ce^{4+}) and Na_2SO_3 can be greatly enhanced by the complex of terbium ion (Tb^{3+}) and the drug with with four emission peaks at 490, 545, 585 and 620 nm [34].

Another method depends on native fluorescence of the drug in 0.1 N H_2SO_4 with λ_{ex} = 290 nm and λ_{em} = 450 nm [35]. One method depends on the quenching effect after binding of the drug to bovine lactoferrin in a dilute aqueous solution with λ_{ex} = 295 nm and emission range from 300 – 550 nm [36].

Flow injection chemiluminescence method depends on reaction of the drug with either cerium and sodium sulphite in acid medium or cerium in acid condition sensitized by rhodamine 6G or luminol- KIO_4 -calcein in alkali medium with emission range from 350 – 550 nm [37].

Another method was based on charge transfer complex formation between lomefloxacin hydrochloride and bromanil with λ_{ex} = 275 nm and λ_{em} = 459 nm [38].

Another process relied on the complex formation of the drug with an aluminum ion to produce a high-fluorescent end product. The amount of sodium dodecyl sulfate added increased the amount of fluorescence that was observed at 429 nm after being excited at 332 nm [39]. The developed method is dependent on the formation of a metal-chelation compound using LMX as a ligand and zinc (II) in an acetate buffer (pH 5.5). The types of metal, their concentrations, pH values, buffer types, and solvents used to dilute them were all

optimized. The best reaction conditions were determined after careful investigation to be 0.2 mM zinc, 2.0 ml acetate buffer (pH 5.5), and water as the diluting solvent. When LMX was excited at 284 nm and then excited at 450 nm, a significant increase in fluorescence intensity was achieved [40].

Also, for the purpose of lomefloxacin detection, a nanocomposite fluorescent probe based on fluorescence quenching was created. The created probe combined the excellent selectivity of molecularly imprinted polymer, the high adsorption affinity of graphene oxide, and the high sensitivity of quantum dots. For monitoring lomefloxacin, the probe demonstrated good sensitivity, high specificity, and rapidity. Lomefloxacin reduced fluorescence emission linearly from 0.10 to 50.0 $\mu\text{g L}^{-1}$, and the probe showed a low limit of detection of 0.07 $\mu\text{g L}^{-1}$ [41].

Lomefloxacin hydrochloride was found in human urine, and a quick, accurate spectrofluorometric method was developed to detect it. The technique is based on the determination of lomefloxacin's native fluorescence in $2 \times 10^{-4} \text{ mol} \cdot \text{L}^{-1}$ at emission = 451 nm following excitation at 323 nm [42].

2.2.3 Electrochemical methods:

This is adsorptive voltametric method depends on using Hg electrode and supporting electrolyte containing Britton-Robinson buffer (pH 8.8) - 0.02 M KCl. The reduction peak of lomefloxacin hydrochloride showed a potential of -1.40 V (vs. Ag/AgCl), [43].

Another method is based on the polarographic catalytic current produced by lomefloxacin hydrochloride in a phosphate buffer (0.125 M) at pH= 6.6 and 2-iodoacetamide solution (2.5×10^{-4} M). The second-order derivative peak current of the catalytic wave of the drug is proportional to its concentration [44].

A differential pulse adsorptive stripping voltametric method using acetate buffer solution (0.04 M) at pH= 4 and accumulation potential of -0.30 V (vs. Ag/AgCl) and accumulation time was 2 minutes [45].

Using a poly-melamine layer modified glassy carbon electrode (p-(melamine)/GCE), a sensitive electrochemical method for the determination of

lomefloxacin has been developed. Horizontal Attenuated Total Reflectance-Infrared Spectroscopy (HATR-IR), Field Emission Scanning Electron Microscopy (FE-SEM), and Electrochemical Impedance Spectroscopy (EIS) were used to characterize the surface morphology of the modified sensor. Square wave voltammetry and cyclic voltammetry were used to measure the electrochemical reactions. The electrode that had been modified with polymer demonstrated excellent electrocatalytic activity in the electrochemical oxidation of lomefloxacin, with a clearly defined voltammetric peak at about 980 mV [46].

This study developed a green direct potentiometric method to measure the antibacterial Lomefloxacin hydrochloride in urine using electrodes that were made in-house. The method uses non-hazardous chemicals and doesn't require sample preparation. The sensor was created using a membrane made of poly vinyl chloride, potassium tetrakis (4-chlorophenyl) borate as a cation exchanger, 2-Nitrophenyloctylether as a plasticizer, and 2-hydroxypropyl--cyclodextrin as a specific molecular recognition component. According to IUPAC recommendations, the proposed sensor was validated, and it displays a linear dynamic range from 1×10^{-5} to $1 \times 10^{-2} \text{ mol} \cdot \text{L}^{-1}$, with a Nernstian slope of 58.914 mV/decade [47].

For the first time, a novel optical sensor for lomefloxacin was based on the plasma resonance characteristics of silver nanoparticles (AgNPs). The change in color and absorption spectra of the AgNPs suspension caused by the hydrogen bonds and electrostatic force between lomefloxacin and AgNPs provided a theoretical foundation for the optical detection of lomefloxacin. Additionally, we increased the sensitivity of the AgNPs-lomefloxacin detection system by adding cystine, which allowed it to reach the critical point of discoloration. Furthermore, it was investigated how the AgNPs-lomefloxacin detection system was affected by variables like temperature, reaction time, and pH 12.

2.2.4 Chromatographic methods:

2.2.4.1 Thin layer chromatographic methods:

The first method depends on densitometric evaluation of thin layer chromatograms of lomefloxacin hydrochloride with its acid degradation product at

288 nm using ammonium chloride solution (0.3 M): n-propanol: conc. ammonia (1:8:1, by volume) as a mobile phase [28].

Another method is based on densitometric evaluation of thin layer chromatograms of lomefloxacin hydrochloride and ciprofloxacin hydrochloride in the presence of their acid induced degradation products at 288 nm using ammonia buffer and methanol (20:80, v/v) as a mobile phase [48]. Also, there is another method that relies on stability indicating densitometric evaluation of thin layer chromatograms of lomefloxacin hydrochloride in the presence of its degradation products at 288 nm

using chloroform: conc. ammonia:methanol (10:3:7, by volume) as a mobile phase [49].

The last method is depending on densitometric evaluation of thin layer chromatograms in bulk drug and tablet dosage form at 288 nm using 2-propanol: conc. ammonia:water (86:6:8, by volume) as a mobile phase [50].

2.2.4.2 High & ultraperformance liquid chromatographic methods:

Table 2: HPLC methods reported for the determination of lomefloxacin.

| Matrix | Column | Mobile phase | Detector | Ref. |
|--|----------------------------------|---|---|--------------------------------------|
| In plasma and urine with norfloxacin (internal standard) | BondaPak C18 | (a) Acetate buffer at pH = 4.8 (b) Acetonitrile (80:23, by volume) | Fluorescence $\lambda_{ex} = 280 \text{ nm}$ $\lambda_{em} = 430 \text{ nm}$ | (Shibl, Tawfik et al. 1991) [51] |
| In plasma with enoxacin (internal standard) | Anion exchange Vydac | acetonitrile: Phosphate buffer at pH = 7 (10:90, by volume) | UV at 280 nm | (Carlucci, Cilli et al. 1993) [52] |
| In plasma with fenbuten and felbinacin | Anion exchange Supelcosil LC-SAX | Acetonitrile Phosphate buffer at pH = 7 | UV at 280 nm | (Carlucci, Mazzeo et al. 1996) [53] |
| In plasma with sarafloxacin (internal standard) | Novapak C18 | Acetonitrile: Phosphate buffer at pH 3 (20:80, v/v) | Fluorescence $\lambda_{ex} = 338 \text{ nm}$ $\lambda_{em} = 425 \text{ nm}$ | (Garcia, Solans et al. 2001) [54] |
| In seminal plasma with ofloxacin (internal standard) | Spherisorb S50DS1-C18 | Acetonitrile: Phosphate buffer at pH 7 (20:80, v/v) | Fluorescence $\lambda_{ex} = 280 \text{ nm}$ $\lambda_{em} = 440 \text{ nm}$ | (Kumar and Goyal 2017) [55] |
| In wastewater with: - Pipemidic acid - Norfloxacin - Ciprofloxacin - Enrofloxacin - Ofloxacin - Sarafloxacin - Difloxacin - Tosufloxacin | YMC ODS-AQ S-3C18 | Gradient elution using: (A) Water (B) Acetonitrile | Tandem mass spectrometry | (Nakata, Kannan et al. 2005) [56] |
| In plasma with ofloxacin | Hibar Lichrospher100 - C8 | 0.5 % triethyl amine at pH = 2.5 with phosphoric acid: acetonitrile (85:15, v/v) | UV at 280 nm | (Zendelovska and Stafilov 2005) [57] |
| In raw material and tablet with excipients | Phenomenex C18 | 1% Acetic acid:methanol:acetonitrile (70:15:15, by volume) | UV at 280 nm | (Tozo and Salgado 2006) [58] |
| In pharmaceutical preparations with: - Gatifloxacin - Levofloxacin - Pefloxacin | LiChrospher 100-C18 | 0.3% of triethylamine at pH = 3.3 with phosphoric acid with acetonitrile:water (20:80, v/v) | UV from 279-295 nm | (Santoro, Kassab et al. 2006) [59] |

| | | | | |
|---|-------------------------------|---|---|--|
| In pharmaceutical preparations with degradation product | Inertsil C18 | Water: acetonitrile: triethyl amine (80:20:0.6, by volume) at pH= 3 with orthophosphoric acid | UV at 328 nm | (Gupta, Yadav et al. 2014) [46] |
| In marine products and animal tissues | Symmetry C18 | Gradient elution using: (a) Acetonitrile (b) 0.1% formic acid at pH =2.5 | Fluorescence $\lambda_{ex} = 280 \text{ nm}$ $\lambda_{em} = 450 \text{ nm}$ | (Chang, Wang et al. 2008) [60] |
| In pharmaceutical preparations with excipients | Inertsil ODS-C18 | Acetonitrile: 0.025 M phosphoric acid (20:80, by volume) | UV at 287 nm. | (Amran, Hossain et al. 2011) [29] |
| In pharmaceutical preparations with enrofloxacin and ofloxacin | Neucleosil C18 | Acetonitrile: Phosphate buffer at pH = 2.4 (20:80, v/v) | UV at 294 nm | (Amin, Dessouki et al. 2011) [61] |
| In pharmaceutical preparations with enrofloxacin and ofloxacin | μ Bondapak C18 | 0.31% ammonium acetate at pH= 2.2 and 0.65% sodium perchlorate and with orthophosphoric acid: acetonitrile (81:19, by volume) | UV at 294 nm | (Amin, Dessouki et al. 2011) [61] |
| A honey sample. With ofloxacin, ciprofloxacin, enrofloxacin, lomefloxacin, and difloxacin. | C18 column | Gradient elution using: (a) methanol (b) acetonitrile (c) 10 mmol/L $\text{NaH}_2\text{PO}_4 \cdot 2\text{H}_2\text{O}$ at PH=3 | fluorescence $\lambda_{em} = 480 \text{ nm}$ | (Tian, Ren et al. 2022) [62] |
| Oxidation of lomefloxacin and balofloxacin | Kinetex 5u XB-C18 100A column | isocratic elution using acetonitrile and 0.05 M phosphate buffer at pH = 3.20 adjusted with o-phosphoric acid (13:87 v/v for lomefloxacin; 20:80 v/v for balofloxacin). | a photodiode array detector | (Żuromska-Witek, Żmudzki et al. 2020) [63] |
| Poultry eggs with enrofloxacin, ciprofloxacin, ofloxacin, pefloxacin, norfloxacin, and sarafloxacin | C18 column | Gradient elution using: (a) 0.1% (V/V) formic acid (b) acetonitrile | Tandem mass spectrometry | (Huang, Fan et al. 2019) [64] |
| Eye drops used in cataract surgery | Phenomenex Luna® C18 column | methanol:water:formic acid (70:29:1, by volume) | Tandem mass spectrometry | (Nassar, Attia et al. 2020) [65] |

2.3 Reported methods of dexamethasone

2.3.1 Spectrophotometric methods:

Dexamethasone sodium phosphate is directly determined at pH = 6 using a double-beam spectrophotometer at 242.5 nm [66] and using water:ethanol (1:2 v/v) as background electrolyte at 240 nm [67]. Oxidation of dexamethasone by iron (III) followed by complexation of iron (II) with potassium hexacyanoferrate (III) to form bluish green complex

with absorbance at wavelength=780 nm is another method [68].

2.3.2 Chromatographic techniques

2.3.2.1 Thin layer chromatographic method

Development of HPTLC method for determination of dexamethasone using hexane– propan-2-ol (90:10, by volume) as a mobile phase [69] (Huetos, Ramos et al. 1999).

2.3.2.2 High and ultraperformance liquid chromatographic methods:

Table 3: HPLC methods reported for the determination of remdesivir.

| Matrix | Column | Mobile phase | Detector | Ref. |
|--|--------------------------------|---|--------------------------------|-------------------------------------|
| In microemulsions | RP-18 column | Water :methanol (35:65; by volume) | UV at 239 nm | (Urban, Mainardes et al. 2009) [70] |
| Pharmaceutical Formulations | C ₁₈ column | Ammonium acetate buffer (5mM) :methanol :acetonitrile(43:25:32, v/v) | UV at 240 nm | (Duarah, Sharma et al. 2021) [71] |
| Pharmaceutical formulations with ofloxacin | C 18 column | Acetonitrile : phosphate buffer at pH= 4 (50:50, v/v) | UV at 236 nm | (Sireesha and Prakash 2012) [72] |
| Pharmaceutical Formulations with moxifloxacin. | BDS Hypersil C8 column | 20 mM phosphate buffer , 0.1% (v/v) triethylamine, atpH = 2.8) and methanol (38.5:61.5 v/v) | diode array detector at 254 nm | (Razzaq, Ashfaq et al. 2017) [73] |
| Pharmaceutical Formulations with granisetron. | CPS Hypersil CN column | Acetonitrile: 100 mM buffer Triethylamine at pH = 3.0 with orthophosphoric acid (25:75 by volume) | UV at 242 nm | (Heda, Kathiriya et al. 2011) [74] |
| Human plasma | a Sphereclone ODS2 column | 10 mM phosphate buffer at pH = 7.0: acetonitrile (68:32,v/v) | UV at 240 nm | (Song, Park et al. 2004) [75] |
| In dried blood spot samples | a Zorbax EclipsePlus C18 colum | Water and acetonitrile with formic acid | Mass spectrometer | (Patel, Tanna et al. 2010) [76] |

3. Conclusion

In this review, simple and clear highlights for the collected information about the most instrumental methods used for determination of compounds used in covid-19 treatment are reported. This review includes HPLC technique, spectroscopy, fluorimetry and electrochemical methods. It was obvious that HPLC technique was the most developed method followed by spectroscopy. The presented methods were validated according to ICH guidelines in addition to some of them were assessed using various green tools.

Conflict of interest statement

Authors declare that there is no conflict of interest.

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