

## SYNTHESIS OF SOME 5-BENZIMIDAZOLE SULPHONAMIDE DERIVATIVES WITH POTENTIAL ANTIMICROBIAL ACTIVITY

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### ABSTRACT

Some new 2-methylbenzimidazole-5-sulphonylchloride derivatives and 2-methylbenzimidazole-5-N-(p-carboxyphenyl)sulphonamide derivatives were synthesized. Some of these compounds were found to possess antimicrobial activities towards different microorganisms.

### INTRODUCTION

The chemical and pharmacological properties of benzimidazole derivatives have commanded the interest of organic and medicinal chemists<sup>(1-4)</sup>. Unfortunately, little is known about 2-methylbenzimidazole-5-sulphonic acid<sup>(5)</sup>. The need for additional information is further magnified by the many useful biological properties e.g. antibacterial (clemizole)<sup>(6,7)</sup>, anthelmintic (cambendazole)<sup>(8,9)</sup>, antitumour [pyrrolo(1,2-a) benzimidazole]<sup>(10-12)</sup>, cardiotoxic (azobenzimidazole)<sup>(13)</sup>, anxiolytic [pyrido(1,2-a)benzimidazole]<sup>(14)</sup> and antiviral agents<sup>(15,16)</sup>.

Furthermore, many substituted sulphonamide derivatives exhibit bacteriostatic, hypoglycemic, diuretic and other pharmacological activities<sup>(17-20)</sup>.

These findings have attracted our attention to preparation of a series of 2-methylbenzimidazole-5-sulphonylchloride derivatives to examine the antimicrobial activities

### EXPERIMENTAL

Thin layer chromatography for analytical purposes were taken on silica gel G glass coated plates and developed with n-butanol-acetic acid-water (4:1:1) using ninhydrin or Dragendorff's reagent as spraying agent. Melting points were uncorrected using Stuart Scientific Melting Point Apparatus SMP2. Infrared spectra were taken in KBr discs using Shimadzu IR 404 spectrophotometer and <sup>1</sup>HNMR spectra were measured using a varian EM 360L Spectrophotometer.

#### A. Synthesis :

2-Methylbenzimidazole I was prepared according to previously reported procedures<sup>(21,22)</sup>.

2-Methylbenzimidazole-5-sulphonic acid II was achieved according to published method<sup>(5)</sup>.

#### 2-Methylbenzimidazole-5-sulphonylchloride III :

This compound was obtained by adding 0.1 mole of 2-methylbenzimidazole to cold 0.2 mole

chlorosulphonic acid at 0-5°C gradually. The reaction mixture was stirred at room temperature for 2h then heated at 60°C for 4h. The mixture was cooled to 5°C. The oily product was washed with cold dioxan several times and left in the refrigerator over night then filtered rapidly. The product was insoluble in 10% Na<sub>2</sub>CO<sub>3</sub> solution; yield 80%, b.p. 96°C (76 mm Hg). IR spectrum (ν, cm<sup>-1</sup>): 3350 (NH), 1370, 1180 (SO<sub>2</sub>).

#### 2-Methylbenzimidazole-5-sulphonamide derivatives IVa-e :

A solution of compound III (0.1 mole) in 10 ml absolute alcohol was added to cold appropriate solution of 0.1 mole of the amino acid (p-aminobenzoic acid, glycine, methionine, alanine and threonine) in 10 ml 1N NaOH solution. The temperature of the reaction mixture was maintained at 0°C until complete addition, then the mixture was heated at 60°C for 3h. The aqueous solution was cooled to 0°C and acidified with 1N HCl to pH 5.3. The crude product was collected and recrystallized from methanol-water mixture (Table 1).

IR spectrum of 2-methylbenzimidazole-5-(N-carboxymethyl)sulphonamide (IVb) (ν, cm<sup>-1</sup>): 3360-2800 (COOH/NH), 1670 (C=O), 1370, 1170 (SO<sub>2</sub>).

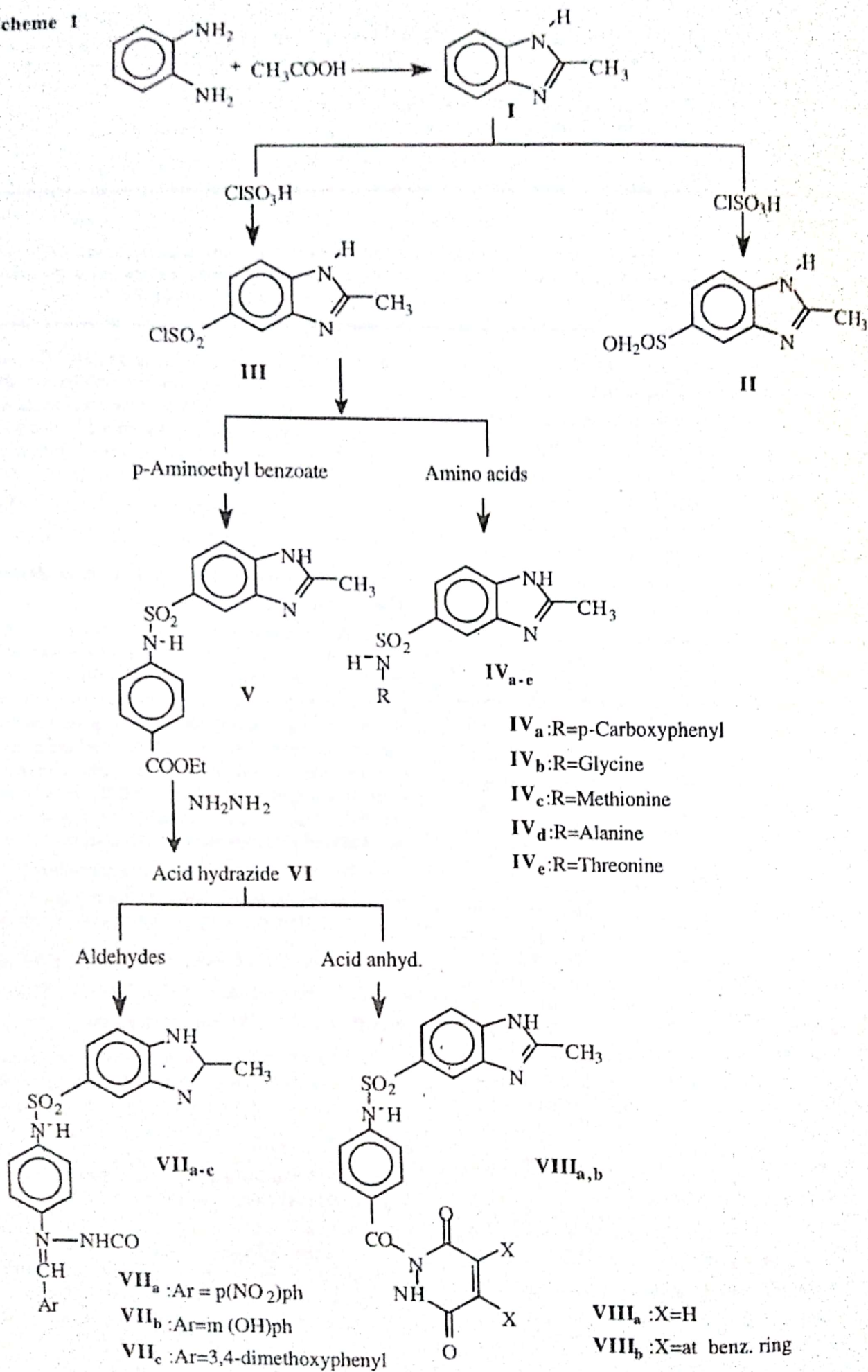
<sup>1</sup>HNMR spectrum of compound IV b (δ, pyridine-d<sub>5</sub>): 2.1 (s, 3H, -CH<sub>3</sub>), 3.3 (s, 2H, -CH<sub>2</sub>), 5.3 (s, NHSO<sub>2</sub>), 7.5-8 (m, ArH), 8.2 (s, 1H, NH benzimidazole).

<sup>1</sup>HNMR spectrum of 2-methylbenzimidazole-5-N-(p-carboxyphenyl)sulphonamide IVa (δ, pyridine-d<sub>5</sub>): 2.2 (s, 3H, -CH<sub>3</sub>), 5.6 (s, br., NHO<sub>2</sub>), 7-8.2 (m, ArH), 8.4 (s, 1H, NH).

#### Unequivocal synthesis of 2-methylbenzimidazole-5-N-(p-carboxyphenyl) sulphonamide (V) :

A solution of 0.1 mole of the compound II in dichloroethane (10 ml) was added dropwise to cold solution of p-aminoethylbenzoate (0.1 mole) in dichloroethane (10 ml) while stirring. The temperature of the reaction mixture was maintained at 5°C until complete addition and then it was refluxed on water

Scheme 1



bath for 7hr. After cooling the product was filtered and recrystallized from methanol (Table 1). IR spectrum ( $\nu$ ,  $\text{cm}^{-1}$ ): 3300 (NH), 1720, 1680 (COOEI), 1370, 1160 ( $\text{SO}_2$ ).  $^1\text{H NMR}$  ( $\delta$ ,  $\text{DMSO-d}_6$ ): 1. (t, 3H,  $\text{CH}_3$  ester) 2.8 (s, 3H,  $\text{CH}_3$ ), 4.2 (q, 2H, J = 3 cps,  $\text{CH}_2$  ester) 6.3 (s, 1H,  $\text{NHSO}_2$ ), 7-8 (m, ArH), 8.3 (s, 1H, NH).

#### 2-Methylbenzimidazole-5-N(9Pcarbohydrazidephenyl) sulphonamide VI :

A mixture of compound V (0.1 mole) and hydrazine hydrate 80% (20 ml) was heated at  $80^\circ\text{C}$  for 9h, then cooled. The product was poured into water then extracted by dichloroethane. The solvent was dried over anhydrous  $\text{Na}_2\text{SO}_4$  and evaporated. An oil product was obtained in yield 84%, b.p  $116^\circ\text{C}$  (76 mm Hg). IR spectrum ( $\nu$ ,  $\text{cm}^{-1}$ ): broad band 3370-3120 (br.  $\text{NHNH}_2$ ), 1670 (C=O), 1370, 1170 ( $\text{SO}_2$ ).

#### 2-Methylbenzimidazole-5-N(p-substitutedbenzylidene carbohydrazidephenyl)sulphonamide (VIIa-c) :

A mixture of 0.1 mole of compound VI and 0.1 mole of the required aldehyde (p-nitrobenzaldehyde, m-hydroxybenzaldehyde or 3,4-dimethoxybenzaldehyde) in 30 ml dichloroethane was heated at  $80^\circ\text{C}$  for 6 h then cooled and filtered. The product was recrystallized from ethanol. (Table 1).

IR spectrum of compound VIIa ( $\nu$ ,  $\text{cm}^{-1}$ ): 3340, 3150 (NH,  $\text{NHSO}_2$ ), 1680 (C=O), 1630 (C=N), 1380, 1170 ( $\text{SO}_2$ ).  $^1\text{H NMR}$  spectrum ( $\sigma$ ,  $\text{DMSO-d}_6$ ): 2.6 (s, 3H,  $\text{CH}_3$ ), 6.8 (s, 1H,  $\text{NHSO}_2$ ), 7.5-8.5 (m, ArH), 8.7 (s, 1H, N = CH), 9.6 (s, 1H,  $\text{NHCO}$ ).

#### 2-Methylbenzimidazole-5-N-(p-(3,6-dioxo-2H-pyridazin-1-yl)carbophenyl)sulphonamide (VIII a-b):

A 0.1 mole of the compound VI was dissolved in 25 ml dichloroethane and 0.1 mole of acid anhydride (maleic or phthalic anhydride) was added while stirring. The mixture was heated at  $80^\circ\text{C}$  for 7h. The solvent was evaporated under vacuum. The product washed with 10%  $\text{Na}_2\text{CO}_3$  solution, then, recrystallized from water. (Table 1).

IR spectrum of compound VIIIb ( $\nu$ ,  $\text{cm}^{-1}$ ): 3350 (NH) 1720, 1680 (C=O), 1370, 1180 ( $\text{SO}_2$ ).  $^1\text{H NMR}$  ( $\delta$ ,  $\text{DMSO-d}_6$ ): 2.6 (s, 3H,  $\text{CH}_3$ ), 6.3 (s, 1H, SONH), 7.5-8.5 (m, ArH), 9.5 (s, 1H, CONH).

#### B- Biological Screening :

The prepared compounds were tested against bacteria (*S. aureus* and *E. coli*) and fungi (*A. niger*) by agar plate diffusion method<sup>(23-25)</sup>. The compounds to be tested were dissolved in 10% aq KOH. Sterile discs of Whatman filter paper containing 15 $\mu\text{l}$  of the above solution were placed over the surface of nutrient agar

containing standardized inoculum from the tested microorganisms.

The plate were incubated at  $37^\circ\text{C}$  for 24h for bacteria and at  $25^\circ\text{C}$  for 48h for fungi.

The inhibition zones appearing after incubation were measured. A disc impregnated with 5 $\mu\text{l}$  of aq. KOH is used as a control. The other disc impregnated with 15 $\mu\text{l}$  of thiabendazole solution in aq. KOH is used as standard for each microorganism. The results are summarized in Table 2.

A main finding is that some of the prepared compounds especially compound II showed a powerfull antifungal activity and compounds IVb, IVc, IVe, VIIa and VIIc showed significant antifungal and antibacterial activities.

#### RESULTS AND DISCUSSION

In the present study, compound I 2-methylbenzimidazole was synthesized according to reported methods<sup>(21,22)</sup>. This compound I was allowed to react with five equivalent chlorosulphonic acid at  $110^\circ\text{C}$  to yield 2-methylbenzimidazole-5-sulphonic acid II.

On the other hand, 2-methylbenzimidazole I was reacted with two equivalent of chlorosulphonic acid at room temperature then at  $60^\circ\text{C}$  to give 2-methylbenzimidazole-5-sulphonylchloride III.

Compound III was then reacted with various amino acids mainly, p-aminobenzoic acid, glycine, methionine, alanine and threonine to afford the corresponding sulphonamide (IVa-e).

Compound IVa was reacted with ethanol and few drops of conc. sulfuric acid to yield 2-methylbenzimidazole-5-N-(p-carbethodzphenyl) sulphonamide V. Also, the compound V was obtained via reaction of 2-methylbenzimidazole-5-sulphonylchloride II with p-aminoethylbenzoate.

Furthermore, through the reaction of compound V with hydrazine hydrate, the acid hydrazide VI was obtained. Compound VI was then allowed to react with some aldehydes mainly P-nitrobenzaldehyde, m-hydroxybenzaldehyde or 3,4-dimethoxybenzaldehyde to afford the corresponding Schiff's bases VIIa-c.

In addition, compound VI was reacted with acid anhydride (maleic anhydride or phthalic anhydride) to give the corresponding pyridazinone derivatives VIII a,b.

The sequence of reactions adopted for the synthesis of the compounds is illustrated in Scheme 1.

The elemental analyses and physical properties of the prepared compounds are listed in Table 1.

The antimicrobial activity of the prepared compounds was examined using agar diffusion method<sup>(23-25)</sup>. The results (Table 2) showed that

Table 1: Melting points, Yield and Elemental Analyses of the Prepared Compounds.

Comp. No.	m.p.C°	yield %	Mol. Formula	Analyses % (Calcd / found)		
				C	H	N
IVa	159	75	C <sub>15</sub> H <sub>13</sub> SN <sub>3</sub> O <sub>4</sub>	54.4	3.9	12.7
IVb	305	65	C <sub>17</sub> H <sub>11</sub> SN <sub>3</sub> O <sub>4</sub>	54.0	4.3	12.2
IVc	260	60	C <sub>13</sub> H <sub>17</sub> S <sub>2</sub> N <sub>3</sub> O <sub>4</sub>	44.6	4.1	15.0
IVd	320	75	C <sub>11</sub> H <sub>13</sub> SN <sub>3</sub> O <sub>4</sub>	45.0	4.5	15.2
IVe	300	63	C <sub>12</sub> H <sub>15</sub> SN <sub>3</sub> O <sub>5</sub>	45.5	4.9	12.2
V	380	76	C <sub>17</sub> H <sub>17</sub> SN <sub>3</sub> O <sub>4</sub>	45.9	5.3	11.8
VI	116	84	C <sub>15</sub> H <sub>15</sub> SN <sub>5</sub> O <sub>4</sub>	46.6	4.6	14.8
VIIa	290	80	C <sub>22</sub> H <sub>18</sub> SN <sub>6</sub> O <sub>5</sub>	46.1	4.2	15.3
VIIb	180	74	C <sub>22</sub> H <sub>19</sub> SN <sub>5</sub> O <sub>4</sub>	46.0	4.8	13.4
VIIc	150	70	C <sub>24</sub> H <sub>23</sub> SN <sub>5</sub> O <sub>5</sub>	46.5	5.2	13.0
VIIIa	320	60	C <sub>19</sub> H <sub>15</sub> SN <sub>5</sub> O <sub>5</sub>	56.8	4.7	11.7
VIIIb	250	68	C <sub>23</sub> H <sub>17</sub> SN <sub>5</sub> O <sub>5</sub>	57.3	4.2	12.0
				49.9	4.2	19.4
				50.3	4.5	19.0
				50.3	4.5	19.0
				55.2	3.8	17.6
				54.8	4.2	18.0
				58.8	4.2	15.6
				59.2	3.9	16.0
				58.4	4.7	14.2
				58.0	4.3	14.5
				53.6	3.5	16.5
				53.2	4.0	16.0
				58.1	3.6	14.6
				58.5	3.3	14.3

Table 2: Antimicrobial Activity of Tested Compounds .

Comp. No	Inhibition Zones (cm)		
	A.niger	E.coli	S.aureus
II	6	-	-
IVb	2.8	5	5
IVc	2.4	4.5	4
IVe	2.2	5	5
VIIa	4	-	4.5
VIIc	3.4	5	4.5
Thiabendazole	7	5	5
Aq.KOH	-	-	-

compound II possess a strong antifungal activity against *A. niger*. Also, the compounds IVb, IVc, IVe, VIIa and VIIc showed a significant activity against the tested organisms (*S. aureus* and *E. coli*).

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## تشبيد بعض مشتقات 0-بنزايמידازول سلفوناميد ودراسة نشاطها المضاد للميكروبات

### نبويه شرف الدين

قسم الكيمياء الصيدلية - كلية الصيدلة - جامعة طنطا - طنطا - مصر

تم تشبيد بعض مشتقات المركب 2-ميثيل بنزايמידازول -0- سلفوناميد ومشتقات المركب 2-ميثيل بنزايמידازول -0- كربوهيدازيد فينيل سلفوناميد.

وقد تم أيضا دراسة تأثير هذه المركبات الشميدة ضد بعض الميكروبات. وقد أظهرت الدراسة التأثير الإيجابي لبعض هذه المركبات ضد أنواع من البكتريا والفطريات.