

## A PHARMACEUTICAL STUDY ON TOPICAL FORMULATIONS OF HENNA

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### ABSTRACT:

The purpose of this study was the preparation of henna extract in a suitable pharmaceutical formulation to be applied topically.

Based on microbiological studies, Hadramout-See'on (dry powder) was selected from number of henna samples collected from different places in Yemen. Hadramout-See'on (dry powder) was extracted using different solvents namely: water, methanol and chloroform separately. The inhibitory activities of various extracts of henna against two species of dermatophytic fungi (*Trichophyton violaceum* and *Microsporum canis*), one species of yeast (*Candida albicans*), one Gram-positive bacteria (*Staphylococcus aureus*) and two Gram-negative bacteria (*Escherichia coli* and *Klebsiella*) were tested. Water extract showed the highest percentage inhibition zone particularly with the tested fungi, followed by methanol extract, while chloroform extract showed weak activity with all strains.

The best active extract of henna was formulated in traditional formulations such as ointments bases (water-soluble base, oleaginous base) and emulsion bases (o/w, w/o emulsion bases), and the in vitro release studies of active component in henna from different formulations were recorded in the following descending order:

Polyethylene glycol base > oil/water (o/w) emulsion base > water/oil (w/o) emulsion base > oleaginous base.

### INTRODUCTION

The first known cultivation of Henna was in Egypt and India, around 5000 B.C. The two main cultivated varieties are *Lawsonia inermis* and *Lawsonia alba* Family: Lythraceae. Henna is widely grown in temperate zones of the world, particularly in Asia and North Africa<sup>(1,2)</sup>.

In Yemen it is now well naturalized in Hadramout, Taiz, Tihama, Aden and Lahej.

Approximately 100 g of henna are needed to stain the hands and feet of an adult, and the dyeing process requires 3-6 hours. The important natural chemical dye ingredient of henna is lawsone (2-hydroxy-1,4-naphthoquinone), constituting about 1% by weight of the crushed leaves<sup>(2,4)</sup>.

Qualitative phytochemical tests, and thin layer chromatography demonstrated the presence of common compounds in the plant extracts including phenols, tannins and flavonoids as major active constituents<sup>(5)</sup>.

Henna is one of the most natural inexpensive, safe drugs in the world, it is safe topically and orally (toxic dose orally is 2.5 g/kg) it is also safe for pregnant woman. Henna might induce hemolysis in G6PD deficient male newborns<sup>(11)</sup>.

Traditionally it is very effective when applied to a first or second degree burns and promotes wound healing, especially chronic wounds and ulcers, this may be due to that henna extracts showed antibacterial activity<sup>(12)</sup>. Henna is not only used for cosmetics, but also for medication as well. For many years, henna has been used for the treatment of skin disorders. A potent fungicide, seborrheic dermatitis and fungal infestations are other reasons why henna is used topically on lesions in some people<sup>(6,9)</sup>. The isolated compound (lawsone) was found to possess significant anti-inflammatory, analgesic, antipyretic activity and anti-tuberculosis activity<sup>(10,13)</sup>.

The aim of this study was the preparation of henna extract in a suitable pharmaceutical formulation to be applied topically.

### MATERIALS AND EQUIPMENTS

#### Materials

Henna samples were collected from different places in Yemen. Sabouraud's agar and Nutrient agar were obtained from Ministry of Health (Yemen). Polyethylene glycol 400, polyethylene glycol 4000, span 80, were obtained from Merck Company (Germany). The following chemicals were of pharmaceutical grade:

Sodium lauryl sulphate (SLS), propylene glycol, liquid paraffin, methanol, chloroform, white soft paraffin, white bees wax, wool fat, borax, and cetyl alcohol were given as a gift from Shaphaco Pharmaceutical Ind. (Sana'a Yemen):

#### Equipment:

UV-Spectrophotometer (Shimadzu U.V-1601 PC, Shimadzu Corporation, Japan).

Electronic Digital Balance (Mettler-Toledo, Ag, CH 8606, Greifensee, Switzerland).

Dissolution apparatus, Erweka, GMBH, D-63150, Type:DT60 (Heusenstamm, Germany)

### METHODS

#### Collection of henna samples

Fresh and dry powder samples of Henna plant were collected from different places in Yemen. Hadramout-See'on and Gaiel Bawazeer, Hajja, Taiz, Haraz, and Tihama.

#### Extraction of henna samples

Each sample (500 gm) was put in a sterile polyethylene bag, sealed and put inside another sealed bag. The samples were then transferred to the laboratory. Plant materials were extracted with methanol (99.6%), chloroform and water separately. Fifty grams of plant material were dissolved in 250 ml of the used solvent (1:5 w/v) in a flask of 500 ml. The flask was shaken for one hour on shaker at 300 r.p.m. The contents of the flask were filtered through a filter paper (Whatman No.1). This procedure was repeated three times on the residue of plant. The obtained solutions of different plant extracts were evaporated to a thick mass by standing on air for a

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Table (1): Effect of different extraction solvents on (Hadramout -See'on dry powder henna extract (5000 ppm) on the growth of different strains.

Extraction Solvent	Percentage of inhibition zone(mm) of different strains						
	Control	<i>Micromporium Canis</i>	<i>Trichophyton violaceum</i>	<i>Candida albicans</i>	<i>Staph. aureus</i>	<i>E.coli</i>	<i>Klebsiella</i>
Water	0	100	100	21	39	35	33
Methanol	0	89	85	18	35	33	32
Chloroform	0	45	42	9	15	12	11

#### Pharmaceutical part

The henna water extract was chosen to be formulated in topical bases, to exert its expected action from different topical preparations. It is important that the vehicle is able to release the active ingredients.

Selection of different topical bases as vehicles for henna extract depends on several factors such as polarity, viscosity, and homogeneity. For this purpose traditional classes of topical bases were investigated which included water-soluble bases, emulsion bases (w/o and o/w emulsions) and oleaginous bases.

As a general rule in ointment formulations if the drug is held firmly by the vehicle, the rate of release of the drug is slow. The release of the drug from ointments can be altered by modifying the composition of the vehicle. A greater release of drug is expected when there is less affinity of the drug for the base<sup>(15)</sup>.

Results illustrated by Fig.1 clearly show that the rate and percentage amount of drug released from polyethylene glycol base is greater than that released from the other bases, in which the percentage of henna released reached 100% within 10 min. The high diffusion rate of henna from water soluble ointment bases that contain mainly polyethylene glycol may be due to diffusion of distilled water through the tea bag and formation of water-PEG solution which increases the solubility and accordingly the rate and extent of henna release. The high release of henna from water-soluble base may be attributed to the high solubility of henna extract in PEG which is water soluble, and the easily permeation of henna from the PEG base through the permeable tea bag to the receptor media similar observations involving the use of PEG base ointment have been made with respect to the release of sorbic acid<sup>(16)</sup>, salicylic acid<sup>(16)</sup>, and benzocaine<sup>(17)</sup>.

Fig. 2: shows that the percentage of the water henna extract released from o/w emulsion base, is higher to some extent than the w/o emulsion base (Fig.3) which could be due to the formation of a continuous contact between the external phase of the o/w emulsion and the distilled water, in which the percentage of henna released reached 80% within 30 min<sup>(14)</sup>.

Alternatively the presence of an oily vehicle as an external phase in w/o emulsion will result in formation of an occlusive film, which will result in a retardation of the permeation of the drug molecules into the sink solution<sup>(14)</sup>.

Fig. 4. shows the release of water henna extract from the oleaginous base. Oleaginous ointment bases contain primarily white soft paraffin with several additional lipoidal constituents which favor the retention of the drug in the base.

Fig.5. shows the amount of water henna extract released from different topical pharmaceutical preparations which can be arranged in the following descending order:

PEG base > o/w emulsion base > w/o emulsion base > oleaginous base.

#### CONCLUSION

From this study it can be concluded that: Water and methanolic extracts of henna have antimicrobial activity on the dermatophytic fungi, gram-positive and gram-negative bacteria, with some priority for water henna extract and it is species dependent.

The amount of henna released from different formulations was arranged in the following descending order:

PEG base > o/w emulsion > w/o emulsion > oleaginous base

Henna water extract is better to be formulated in PEG (water-soluble) bases and o/w emulsion bases respectively.

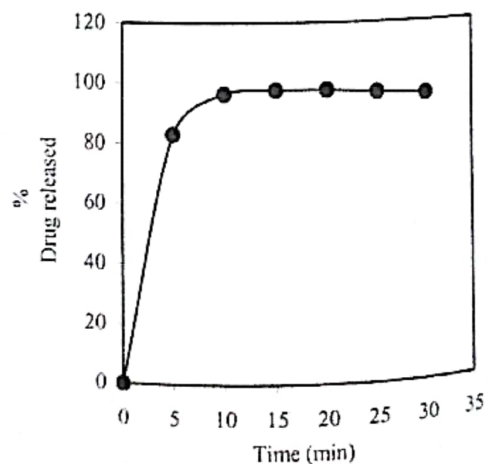


Fig. 1: In vitro drug release from polyethylene glycol base into distilled water at 37°C.

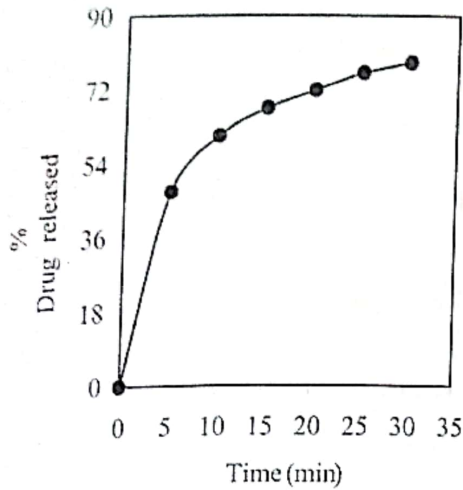


Fig. 2: *In vitro* drug release from o/w emulsion base into distilled water at 37°C.

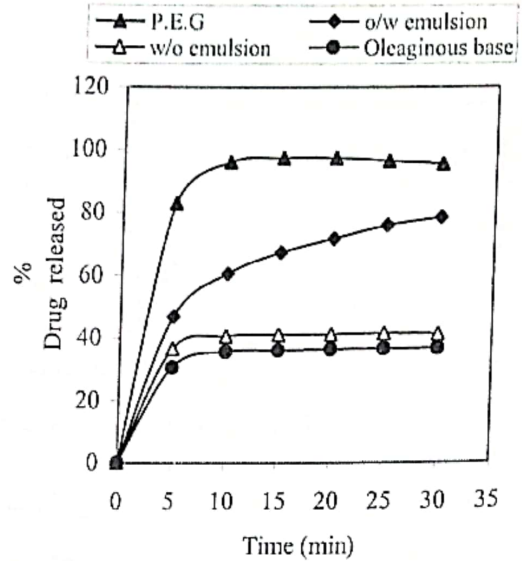


Fig. 5: *In vitro* release of henna from different ointment bases into distilled water at 37°C.

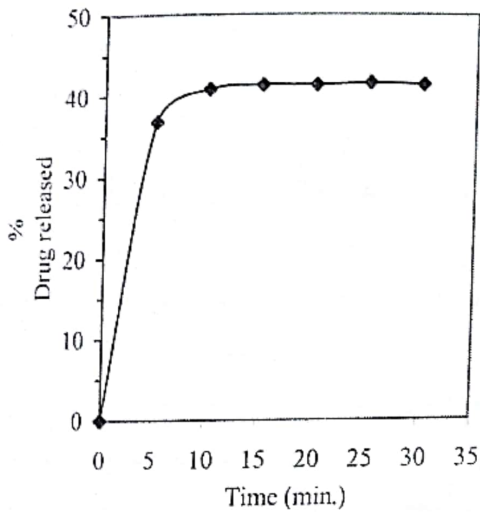


Fig. 3: *In vitro* drug release from w/o base into distilled water at 37°C.

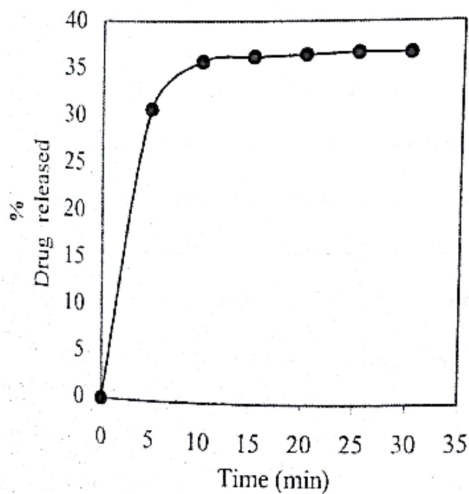


Fig. 4: *In vitro* drug release from oleaginous base into distilled water at 37°C.

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## دراسة صيدلانية لصياغات موضعية من الحناء.

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الغرض من هذه الدراسة هو الوصول لأفضل مستحضر صيدلاني موضعي لمستخلص الحناء، والذي جمع من عدة مناطق مختلفة في الجمهورية اليمنية، وذلك بعد دراسة فعالية المستخلصات المختلفة على ثلاثة أنواع من الفطريات المسببة للأمراض الجلدية ونوع واحد من الخمائر، ونوع واحد من البكتريا الموجبة الجرام واثنين من البكتريا السالبة الجرام. وينقسم هذا البحث إلى جزئين:

الجزء الأول:

ويشمل استخلاص المادة الفعالة في الحناء ودراسة التأثير الفطري والبكتيري لأحسن مستخلص: تم الاستخلاص بعدة مذيبات حيث أظهر كل من المستخلص المائي والميثانولي للحناء تأثيراً مثبطاً بدرجة عالية للفطريات والبكتريا المختبرة، وذلك مع إظهار بعض التفوق للمستخلص المائي، في حين أظهرت خلاصة الكلوروفورم تأثيراً ضعيفاً على كل من الفطريات والبكتريا المختبرة. كذلك وجد أن مستخلص الحناء من المسحوق الجاف أكثر فعالية من مستخلص الأوراق الخضراء. كما وجد أن ترتيب الفعالية للمستخلص المائي من المسحوق الجاف للحناء والذي جمع من عدة مناطق مختلفة في اليمن كالآتي:

حضر موت (سيون) < حضر موت (غيل باوزير) < حجة < تعز < حراز < تهامة.

الجزء الثاني:

ويشمل تحضير أفضل مستخلص من الحناء في عدة قواعد موضعية صيدلانية مختلفة ودراسة انطلاق المادة الفعالة من الحناء:

تم اختيار المستخلص الأكثر فعالية (للمستخلص المائي من المسحوق الجاف - حضر موت سيون) ليكون موضع الدراسة حيث تم صياغته في عدة صيغ صيدلانية موضعية مختلفة من المراهم (مرهم يذوب في الماء ومرهم دهني) والمستحلبات (زيت في الماء، ماء في الزيت).

تمت دراسة انطلاق الحناء من القواعد الموضعية المختلفة وفي أوقات متدرجة في جهاز معد لذلك وصولاً إلى حساب النسبة المئوية المنطلقة من الحناء في كل وقت و لكل قاعدة على حدة حيث أظهرت النتائج لانطلاق مادة الحناء من الصيغ الموضعية المحضرة الترتيب التنازلي الآتي:

قاعدة عديد اثيلين جليكول < مستحلب الزيت في الماء < مستحلب الماء في الزيت < القاعدة الدهنية